Lab:	The	<b>Briny</b>	Deep
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Name: \_\_\_\_\_ Period: \_\_\_\_\_

## Introduction

Although not as visible as tides and waves, large movements of ocean water called upwellings, surface currents, and density currents have a powerful influence on our planet. Weather, climate, fishing, and animal migrations are all impacted by these circulation patterns. This activity will focus on the type of circulation known as "density currents".

## **Pre-Lab Questions**

1. Match each type of current with its definition.

\_\_\_\_ Upwelling A. a continuous flow of water along a constant path in the ocean

ocea

B. the movement of cold, deep water to the ocean surface

Density Currents

Surface Currents

C. the sinking of water that has become heavier than the

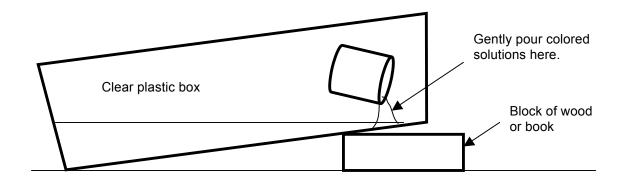
surrounding water.

2. List the following materials from highest to lowest density: aluminum, gold, water, propane, lead, oxygen, mercury, and vegetable oil.

3. Everyone is familiar with the symbol  $(^{0}/_{0})$ , which means percent, or parts per \_\_\_\_\_\_. A similar symbol,  $(^{0}/_{00})$ , is used to indicate how salty a sample of water is. What does this symbol mean?

It means parts per \_\_\_\_\_\_.

4. The salinity of Atlantic Ocean Water is  $35^{0}/_{00}$ . Therefore, 1000 g. of this water would include g. of water and g. of salt.



## **Instructions**

1.	Obtain a clear plastic box. Add 800 ml of room temperature water to the box. Use two hands to
	carry it back to your table because the boxes crack really easily. Set up the box as shown on the
	front of this sheet. You will also need a small beaker and a spoon. Let the water become calm,
	and try not to disturb it throughout the lab.

2.	Put 30-40 ml of room temperature water into the small beaker.	<ul> <li>Add about half a spoon of salt and</li> </ul>
	2 drops of yellow food coloring. Stir the water until the salt dis	solves.

	2 dropo of your food coloring. Our the water until the call discorrect.					
3.	<b>Carefully and slowly</b> pour the solution into the clear plastic box near the end that is propped up. Get down and look into the side of the box. Describe what happened.					
4.	Next, get 30-40 ml of really cold water. Add 2 drops of blue food coloring and stir. Do not pour it into the box yet. Predict what will happen when you pour it in.					
5.	5. Carefully and slowly pour the blue, cold water in. Describe what happened.					
6.	Refill your beaker with 30-40 ml of hot water. Add 2 drops of red food coloring. Predict what w happen when this is poured in.					
7.	Carefully and slowly pour the hot water in. Describe what happened.					
8.	From what you have learned so far, which has a higher density? Circle one.					
	salty water freshwater					
9.	From what you've learned so far, which has a higher density? Circle one.					
	warm water cold water					
10	. Use colored pencils (yellow, blue, red) to color the diagram on the front of this sheet to show what the layers looked like <u>after the hot, red water was poured in</u> .					

11. Use **two hands** to carry your box of water to a sink. Dump it out. Rinse your small beaker and spoon off and put them back. Answer the follow-up questions on the next page.

Use sentences to answer every question marked with an asterisk (\*).

1. *Hov	v does evaporation from	an ocean increa	ise the salinity of	f the water that didn't eva	aporate? (p. 424)				
2. Why	does the density of liqui	id water increase	e as it cools? Cir	rcle one.					
	A. The molecules slow down, so they are packed closer together.								
	B. The molecules slow down, so they are spread farther apart.								
	ok at figure 5 on p. 452 a freezes?	and read the cap	tion. Why does	water beneath the surfac	e get saltier as water on the				
4. <u>Read</u> one.	d p. 452 and look at figu	re 6. What caus	es the Mediterra	nean Sea to be so salty	compared to the Atlantic? Circle				
one.	lots of rain	lots of evapora	tion	formation of ice	ice/snow melting				
5. *Wh <u>։</u>	y is Antarctic Bottom Wa	ater (ABW) more	dense than othe	er layers in the Atlantic?	(Hint: It is not very salty.)				
6. Wou	ld it be easier for you to	float in the Atlan	tic or the Medite	rranean?					
7. Look	at the graph on p. 424.	At which latitude	e is sea surface	temperature the warmes	t?				
8. Why	is the salinity of water n	ear the equator	fairly low? (Rea	nd top paragraphs on p. 4	24.) Circle one.				
	lots of rain	lots of evapora	tion	formation of ice	ice/snow melting				
9. Why	is the salinity fairly high	around 30 degre	ees north and so	uth of the equator?*					
	lots of rain	lots of evapora	tion	formation of ice	ice/snow melting				
10. *WI	hy is the salinity of wate	r near the poles I	nigher in the win	ter and lower in the sumr	mer? (p. 424)				
	e movement of cold wating to happen? (p. 450)	ter from the ocea	n depths to the s	surface is called "upwellii	ng". Explain how wind causes				
	gions of upwelling are ty o be good fishing areas?			areas. What does upwell	ing bring to the surface, causing				
	algae	fish	oxygen	nutrients	plankton				